This article was downloaded by:

On: 24 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



Journal of Liquid Chromatography & Related Technologies

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713597273

TOPOLOGICAL INDEXES FOR EVALUATION OF THE SEPARATION OF D AND L AMINO ACIDS BY TLC>

A. Pyka^a

^a Faculty of Pharmacy, Academy of Medicine, Sosnowiec, Poland

Online publication date: 13 January 2005

To cite this Article Pyka, A.(1999) 'TOPOLOGICAL INDEXES FOR EVALUATION OF THE SEPARATION OF D AND L AMINO ACIDS BY TLC>', Journal of Liquid Chromatography & Related Technologies, 22: 1, 41 - 50

To link to this Article: DOI: 10.1081/JLC-100101642 URL: http://dx.doi.org/10.1081/JLC-100101642

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

TOPOLOGICAL INDEXES FOR EVALUATION OF THE SEPARATION OF D AND L AMINO ACIDS BY TLC

A. Pyka

Faculty of Pharmacy Academy of Medicine 4, Jagiellońska Street PL-41-200 Sosnowiec, Poland

ABSTRACT

Selected amino acids have been separated into their enantiomers by thin layer chromatography on a chiral stationary phase. A new valence optical topological index (I_{opt}^{v}) and valence optical Gutman index (M_{2opt}^{v}) are proposed which enable distinction between isomers of L and D configuration.

INTRODUCTION

Thin layer, gas, and liquid chromatographic methods have been developed to separate amino acids enantiomers and their derivatives. Analysis of amino acids and their derivatives have a great significance in medical diagnosis of hereditary metabolic diseases (phenylketonuria and other aminoaciduria). My previous investigations have concerned a new optical topological index (I_{opt}) for predicting the separation of selected D and L optical isomers by TLC.

The aim of the work reported herein was to propose new optical topological indexes and to evaluate the correlation between these and other topological indexes, and the R_F and R_M values of the optical isomers of D and L configuration of amino acids separated by TLC on a chiral stationary phase.

EXPERIMENTAL

Thin Layer Chromatography

The optical isomers of D and L amino acids: alanine, norvaline, norleucine, glutamic acid, phenylalanine, tyrosine, tryptophan, and proline were separated on Chir HPTLC plates (E. Merck, Darmstadt, Germany). Before use, plates were activated for 15 min at 100°C. Amino acids were separated using methanol-water-acetonitrile (1:1:4) as the mobile phase. The development distance was 11 cm. The chromatograms were visualized by dipping into a solution of ninhydrin (0.3%) in acetone and heating at 110°C for 10 min.

Constant of the Pair Separation R_F^{α}

The constant of the pair separation (R_F^{α}) were calculated for the investigated pairs of optical isomers as the ratio of the R_F value of the L isomer to R_F value of the D isomer.¹⁷

Calculation of Topological Indexes Described in the Literature

The Randić $(^0\chi^{\nu})$, 18 Wiener (W), 19,20 and optical topological $(I_{opt})^{13}$ indexes were calculated for the investigated amino acids.

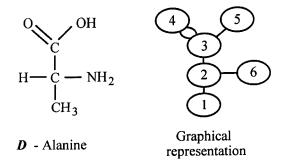
Calculation of Topological Indexes Proposed in this Work

Valence optical topological index (I_{opt}^{v})

Definition of valence optical topological index

Let *G* be a molecular graph and $x_1, x_2, ..., x_n$ its vertexes. ^{21,22} Let $Z^{v} = \|Z_i^{v}\|$

be an n-dimensional column-vector, such that its i-th component corresponds to the vertex of the molecular graph G. For all values of i, i=1, 2, ..., n, we choose



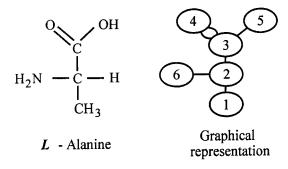


Figure 1. Graphical representation of the structures of *D* and *L* alanine.

 $Z_i^{\nu} = \delta_i$, except if the *i*-th vertex of G represents an asymmetric (optically active) carbon atom. In this latter case, $Z_i^{\nu} = \delta_i$ if the respective carbon atom has configuration D, and $Z_i^{\nu} = -\delta_i$ if the configuration of that carbon atom is L.

The degree of the i-th vertex of the (hydrogen-suppressed) molecular graph is calculated as

$$\delta_{\mathbf{i}} = \zeta_{\mathbf{i}} - h_{\mathbf{i}} \tag{1}$$

where ζ_i is the number of valence electrons of the *i*-th atom in the molecule considered, and h_i is the number of hydrogen atoms attached to that atom.

Let
$$S'^{\nu} = \|S_i^{\nu}\|$$
 be another column-vector calculated as
$$S'^{\nu} = [D] \bullet Z^{\nu}$$
 (2)

where [D] is distance matrix of graph $G_{*}^{18,20,22}$ and \bullet indicates usual matrix multiplication.

Then, the valence optical topological index is given by

$$I_{\text{opt}}^{\nu} = \sum_{i=1}^{n} S_{i}^{\nu} \tag{3}$$

Calculation of I_{opt}^{v} for D- and L-alanine

Firstly, graphical representations of the structures of D- and L-alanine, with numbered vertexes, are established as in Fig. 1. The distance matrix, [D], is then constructed from these graphical representations, the elements being determined by means of the values given by Barysz et al. 20 (see Table 1).

The distance matrixes for the D & L amino acids are identical.

The distance matrix constructed above is multiplied by column vector Z_D^{ν} or Z_L^{ν} to give the column vector $\mathbf{S}^{\prime \nu}$ which consists of elements $S_i^{\prime \nu}$.

 $[\mathbf{D}] \times \mathbf{Z}_{\mathrm{D}}^{\nu}$ for *D*-alanine is given by:

			[D]			7	$Z_{\mathrm{D}}^{\mathrm{v}}$		S_{D}^{ν}	
Γ 0	1	2	2,375	2,750	1,857		[1]		44,571	ĺ
1	0	1	1,375	1,750	0,857		3		24,571	
2	1	0	0,375	0,750	1,857		4		16,571	
2,375	1,375	0,375	0,250	1,125	2,232	•	6	=	21,821	
2,750		0,75		0,250	2,607		5		26,821	
1,857	0,857	1,857	2,232	2,607	0,143		3		38,712	

Therefore, I_{opt}^{ν} for *D*-alanine is calculated as follows:

$$I_{opt}^{v} = 44,\!571 +\!24,\!571 +\!16,\!571 +\!21,\!821 +\!26,\!821 +\!38,\!712 = 173.067$$

Vertex Number	1	2	3	4	5	6
1	0	1,000	2,000	2,375	2,750	1,857
2	1,000	0	1,000	1,375	1,750	0,857
3	2,000	1,000	0	0,375	0,750	1,857
4	2,375	1,375	0,375	0,25	1,125	2,232
5	2,750	1,750	0,750	1,125	0,25	2,607
6	1,857	0,857	1,857	2,232	2,607	0,143

 $[\mathbf{D}] \times Z_L^{\nu}$ for *L*-alanine is given by:

			[D]				Z_D^{ν}		$S_D^{'\nu}$	
0	1	2	2,375	2,750	1,857		1		[38,571]	
1	0	1	1,375	1,750	0,857		-3		24,571	
2	1	0	0,375	0,750	1,857		4	_	10,571	
2,375	1,375	0,375	0,250	1,125	2,232	•	6	=	13,571	
2,750	1,75	0,75	1,125	0,250	2,607		5		16,321	
1,857	0,857	1,857	2,232	2,607	0,143		3		33,570	

A calculation analogous to that for D-alanine shows that, for L-alanine, $\rm I^{v}_{opt}=137.175$

$\textit{Valence optical Gutman index} \ (\ M_{\ 2opt}^{\, \nu})$

A valence optical Gutman index, $\,M_{\,2\text{opt}}^{\,\nu}\,,$ similar to the Gutman index $M_{\,2},$ is given by

$$M_{2\text{opt}}^{\nu} = \sum_{\text{all edges}} (\delta_{i} \delta_{j}) \tag{4}$$

where δ_i is calculated by Eq. (1); a value of " $+\delta_i$ " is assigned to an asymmetric carbon atom bound to an amino group which gives the molecule the D configuration; a value of " $-\delta_i$ " is assigned to an asymmetric carbon atom bound to an amino group which gives the molecule the L configuration.

Calculation of M^v_{2opt} for D- and L-alanine

For **D**- alanine:

$$M_{2opt}^{v} = (1\times3) + (3\times4) + (3\times3) + 2(4\times6) + (4\times5) = 92$$

For *L*- alanine:

$$M_{2opt}^{v} = [1 \times (-3)] + [(-3) \times 4] + [(-3) \times 3] + 2(4 \times 6) + (4 \times 5) = 44$$

RESULTS AND DISCUSSION

 R_F and R_M values; numerical values of the topological indexes W, ${}^0\chi^{\nu}$, and I_{opt} ; and numerical values of the new topological indexes I^{ν}_{opt} and M^{ν}_{2opt} calculated using the procedures described above are listed in Table 2 for the D and L amino acids investigated.

Numerical values of the topological indexes W and ${}^0\chi^{\rm v}$ are equal for the pair of isomers, i.e., these indexes do not differentiate L and D isomers. In this work, applied *optical topological index* ($I_{\rm opt}$), *valence optical topological index* ($I_{\rm opt}^{\rm v}$), and *valence optical Gutman index* ($M_{\rm 2opt}^{\rm v}$) enable the differentation of optical isomers of L and D configuration.

Constants of the pair separation, R_F^{α} , are listed in Table 3 for the L and D amino acids investigated.

 R_F^{α} values characterizes the separation of isomers pairs of L and D configuration. A similar method for evaluation of pair separations was applied by J. Śliwiok.¹⁷

Table~2 Numerical Values of R_F and $R_M,$ and the Topological Indexes ~W, $^0\!\chi^\nu~,~I_{opt}~,~i_{opt}~,~and~M^\nu_{2opt}$

Amino Acid	Retention Data*		Topological Indexes						
	R _F	R _M	W	°χ ^ν	I _{opt}	I^{ν}_{opt}	M v 2opt		
D-Alanine L-Alanine	0.47	0.052	24.553	3.510	48.463	173.067	92		
	0.53	-0.052	24.553	3.510	36.499	137.175	44		
<i>D</i> -Norvaline <i>L</i> -Norvaline	0.50	0.000	63.517	4.924	126.391	398.137	101		
	0.56	-0.105	63.517	4.924	104.427	332.245	47		
D-Norleucine L-Norleucine	0.51	-0.017	94.499	5.631	188.355	571.672	105		
	0.60	-0.176	94.499	5.631	158.391	481.780	51		
D-Glutamic acid L-Glutamic acid	0.52	-0.035	120.106	5.280	239.069	1000.777	175		
	0.58	-0.140	120.106	5.280	202.855	892.135	121		
D-Phenylalanine L-Phenylalanine	0.45	0.087	175.472	6.604	350.301	1261.313	163		
	0.54	-0.070	175.472	6.604	302.331	1117.403	109		
D-Tyrosine L-Tyrosine	0.55	-0.087	218.713	6.974	436.533	1677.363	189		
	0.65	-0.269	218.713	6.974	379.061	1504.947	135		
D-Tryptophan L -Tryptophan	0.50	0.000	309.209	8.104	617.632	2272.183	238		
	0.58	-0.140	309.209	8.104	541.948	2045.131	184		
D-Proline L-Proline	0.38	0.213	54.373	4.554	108.103	379.950	114		
	0.46	0.070	54.373	4.554	88.771	321.774	54		

^{*} Average of 8 measurements

Numerical values of the indexes $I_{opt(L/D)}$ $(I_{opt(L/D)} = \frac{I_{opt(L)}}{I_{opt(D)}})$ and $I_{opt(L/D)}^{\nu}$ $(I_{opt(L/D)}^{\nu} = \frac{I_{opt(L)}^{\nu}}{I_{opt(D)}^{\nu}})$ for optical isomers of L and D configuration can used to differentiate these isomers on a chromatographic and optical basis. Numerical values of the indexes $I_{opt(L/D)}$ and $I_{opt(L/D)}^{\nu}$ are comparable with numerical values of constants of the pair separation R_F^{α} for L and D isomers. However, the ratio of the M_{2opt}^{ν} value of the L isomer to the M_{2opt}^{ν} value of the

 $\label{eq:Table 3}$ Numerical Values of R_F^α , $I_{opt(L/D)},~I_{opt(L/D)}^\nu$, and $M_{2opt(L/D)}^\nu$

Amino Acid	R_F^{α} $= \frac{R_{F(L)}}{R_{F(D)}}$	$I_{opt(L/D)} = \frac{I_{opt(L)}}{I_{opt(D)}}$	$I_{opt(L/D)}^{\pmb{\nu}}$	$M_{2\text{opt}(L/D)}^{\text{V}} = \frac{M_{2\text{opt}(L)}^{\text{V}}}{M_{2\text{opt}(D)}^{\text{V}}}$
Alanine	1.128	0.753	0.793	0.478
Norvaline	1.120	0.826	0.834	0.465
Norleucine	1.176	0.841	0.843	0.486
Glutamic acid	1.115	0.849	0.891	0.691
Phenylalanine	1.200	0.863	0.886	0.669
Tyrosine	1.182	0.868	0.897	0.714
Tryptophan	1.160	0.877	0.900	0.773
Proline	1.210	0.821	0.847	0.474

D isomer have different numerical values for different isomer pairs. The obtained values for the *optical topological indexes* and *valence optical topological indexes* have high repeatability that range from 0.753 to 0.877 and from 0.793 to 0.900 for the indexes $I_{opt(L/D)}$ and $I_{opt(L/D)}^{\nu}$, respectively. This indicates an existing regularity between the chemical consitution that is expressed by the *optical coefficients of distinction* $(I_{opt(L/D)})$ and $I_{opt(L/D)}^{\nu}$ and the chromatographic data that is obtained as the ratio of the $R_{F(L)}$ and $R_{F(D)}$ values (the constants of the isomer pairs R_F^{α}) for the L and D amino acids investigated.

REFERENCES

- 1. H. Frank, G. J. Nicholson, E. Bayer. J. Chromatogr. Sci., 15, 174-176 (1977).
- 2. W. H. Pirkle, T. C. Pochapsky. J. Am. Chem. Soc., 108, 352-374 (1986).
- 3. S. Lam. J. Chromatogr., 355, 157-164 (1986).
- 4. J. Martens, R. Bhushan. Int. J. Peptide Protein Res., 34, 433-444 (1989).
- 5. Chromatographic Chiral Separations. M. Zief, L. J. Crane, eds. Chromatographic Science Series Vol. 40. New York, Basel: Marcel Dekker, Inc. (1988).

- R. Bushan, J. Martens. "Amino Acids and Their Derivatives," in *Handbook of Thin-Layer Chromatography*, J. Sherma, B. Fried, eds. *Chromatographic Science Series*, Vol. 71. New York: Marcel Dekker, Inc. pp. 389-425.
- 7. A. Alak, D. W. Armstrong. *Anal. Chem.*, **58**, 582-584 (1986).
- 8. K. Günther, M. Schickedanz. Naturwissenchaften, 72S, 149-150 (1985).
- 9. R. Bhushan, I. Ali. Chromatographia, 23, 141-142 (1987).
- 10. R. Bhushan, I. Ali. J. Chromatogr., 392, 460-563 (1987).
- 11. J. Sherma. J. Planar Chromatogr. Mod. TLC, 10, 320-331 (1997).
- 12. I. I. Malakhova, B. V. Tiaglov, E. V. Degtiarev, V. D. Krasikov. 8th International Symposium on Instrumental Planar Chromatography, 1995. Interlaken, Switzerland, poster, Q9.
- 13. A. Pyka. J. Planar Chromatogr.-Mod. TLC, 6, 282-288 (1993).
- 14. U. A. Th. Brinkman, D. Kamminga. J. Chromatogr., 330, 375-378 (1985).
- 15. K. Günther. J. Chromatogr., 448, 11-30 (1988).
- K. Günther, K. Möller. "Separation of Enantiomers by Thin Layer Chromatography," in *Handbook of Thin-Layer Chromatography*,
 J. Sherma, B. Fried, eds. *Chromatographic Science Series*, Vol. 71. New York: Marcel Dekker, Inc. pp. 389-425.
- 17. J. Śliwiok, Z. Kwapniewski. *Pedagogical University in Katowice, Section of Chemistry*, **4**, 47-50 (1963). (in Polish)
- 18. N. Trinajstić. *Chemical Graph Theory*, Boca Raton, Florida: CRC Press (1992).
- 19. H. Wiener. J. Am. Chem. Soc., 69, 17-20 (1947).
- 20. M. Barysz, G. Jashari, R. S. Lall, V. K. Srivastava, N. Trinajstić. "Chemical Applications of Topology and Graph Theory," *Studies in Physical and Theoretical Chemistry*, **28**, 222-230 (1983).

21. I. Gutman, O. E. Polansky. *Mathematical Concepts in Organic Chemistry*. Berlin: Springer-Verlag (1986).

22. I. Gutman, A. Pyka. J. Serb. Chem. Soc., 62, 261-265 (1997).

Received May 19, 1998 Accepted June 15, 1998 Manuscript 4821-TLC

Request Permission or Order Reprints Instantly!

Interested in copying and sharing this article? In most cases, U.S. Copyright Law requires that you get permission from the article's rightsholder before using copyrighted content.

All information and materials found in this article, including but not limited to text, trademarks, patents, logos, graphics and images (the "Materials"), are the copyrighted works and other forms of intellectual property of Marcel Dekker, Inc., or its licensors. All rights not expressly granted are reserved.

Get permission to lawfully reproduce and distribute the Materials or order reprints quickly and painlessly. Simply click on the "Request Permission/Reprints Here" link below and follow the instructions. Visit the U.S. Copyright Office for information on Fair Use limitations of U.S. copyright law. Please refer to The Association of American Publishers' (AAP) website for guidelines on Fair Use in the Classroom.

The Materials are for your personal use only and cannot be reformatted, reposted, resold or distributed by electronic means or otherwise without permission from Marcel Dekker, Inc. Marcel Dekker, Inc. grants you the limited right to display the Materials only on your personal computer or personal wireless device, and to copy and download single copies of such Materials provided that any copyright, trademark or other notice appearing on such Materials is also retained by, displayed, copied or downloaded as part of the Materials and is not removed or obscured, and provided you do not edit, modify, alter or enhance the Materials. Please refer to our Website User Agreement for more details.

Order now!

Reprints of this article can also be ordered at http://www.dekker.com/servlet/product/DOI/101081JLC100101642